

A FLAVANONE FROM *LANNEA ACIDA*

SARWAT SULTANA and M. ILYAS

Department of Chemistry, Aligarh Muslim University, Aligarh 202001 (U.P.), India

(Revised received 20 August 1985)

Key Word Index—*Lannea acida*; Anacardiaceae; 6,7,2,2-dimethylchromeno-8, γ,γ -dimethylallyl flavanone.

Abstract—Chromatographic separation of the acetone extract of *Lannea acida* leaves resulted in the separation of a new flavanone, 6,7,2,2-dimethylchromeno-8, γ,γ -dimethylallyl flavanone.

INTRODUCTION

Lannea acida is a small deciduous tree. It is an important drug in the indigenous system of treatment in North Nigeria. The leaves are used as a febrifuge and have been described to be useful in the treatment of gout and rheumatism [1]. No work has been reported on this medicinally important plant. In this communication, we deal with the isolation of a new prenylated chromenoflavanone.

RESULTS AND DISCUSSION

Column chromatography of the acetone extract of the leaves and crystallization yielded (LA-1) white needles, $C_{25}H_{26}O_3$, mp 91–92°. The UV maxima at 222 and 272 nm classified it as an isoflavone or a flavanone [2]. With aqueous NaOH and conc. H_2SO_4 it gave orange-red and dark red colours, respectively, characteristic only of a flavanone. It is readily cleaved with hot aqueous NaOH to give a yellow chalcone, mp 113–115° [3].

The 1H NMR spectrum of the compound showed evidence for the presence of a γ,γ -dimethylallyl unit and a 2,2-dimethylchromene ring. The singlet at δ 1.46 for six protons and the two doublets at δ 5.6 and 6.7 ($J = 11$ Hz) indicated a dimethylchromene ring. The two three proton singlets at δ 1.75 and 1.80 and the doublet at δ 3.2 ($J = 7$ Hz) for two protons and an olefinic proton triplet at δ 5.4 ($J = 7$ Hz) revealed the presence of a γ,γ -dimethylallyl unit. Besides these a sharp singlet at δ 7.5 for five protons (unsubstituted phenyl) and the singlet at δ 7.7 for the C-5 proton suggested that the compound has a trisubstituted 'A' ring having the chromene as well as the dimethylallyl unit and the unsubstituted side phenyl. The presence of a two protons multiplet at ca δ 3.0 and a one proton multiplet at ca δ 5.15 clearly defined the compound as a flavanone. The NMR spectra of several prenylated chromeno-flavanoids have been reported [3–5] and the above values are in good agreement with them.

The mass spectrum of the flavanone showed an $[M]^+$ peak at m/z 374 (10%). Scheme 1 outlines the main pathways of fragmentation. The peak at m/z 319 could be accounted for by the loss of C_4H_7 . This is analogous to the

cleavage of the side chain in scandenone-4-methyl-ether [6].

The above results and biogenetic considerations [7] enable us to formulate the compound LA-1, as 6,7,2,2-dimethylchromeno-8, γ,γ -dimethylallyl flavanone.

EXPERIMENTAL

1H NMR spectra were recorded at 100 MHz in $CDCl_3$ and $DMSO-d_6$ (TMS int. standard) and values are given in ppm. MS were recorded using a direct inlet system. The progress of CC (silica gel, Merck) was monitored by TLC on Merck silica gel HF_{254} plates visualized by exposure to UV light.

Plant material. Fresh leaves of *L. acida* (Rich) were collected from the campus of Ahmedo Bello University, Zaria, Nigeria. A voucher specimen has been deposited in the Herbarium of the Botany Department, Aligarh Muslim University, Aligarh.

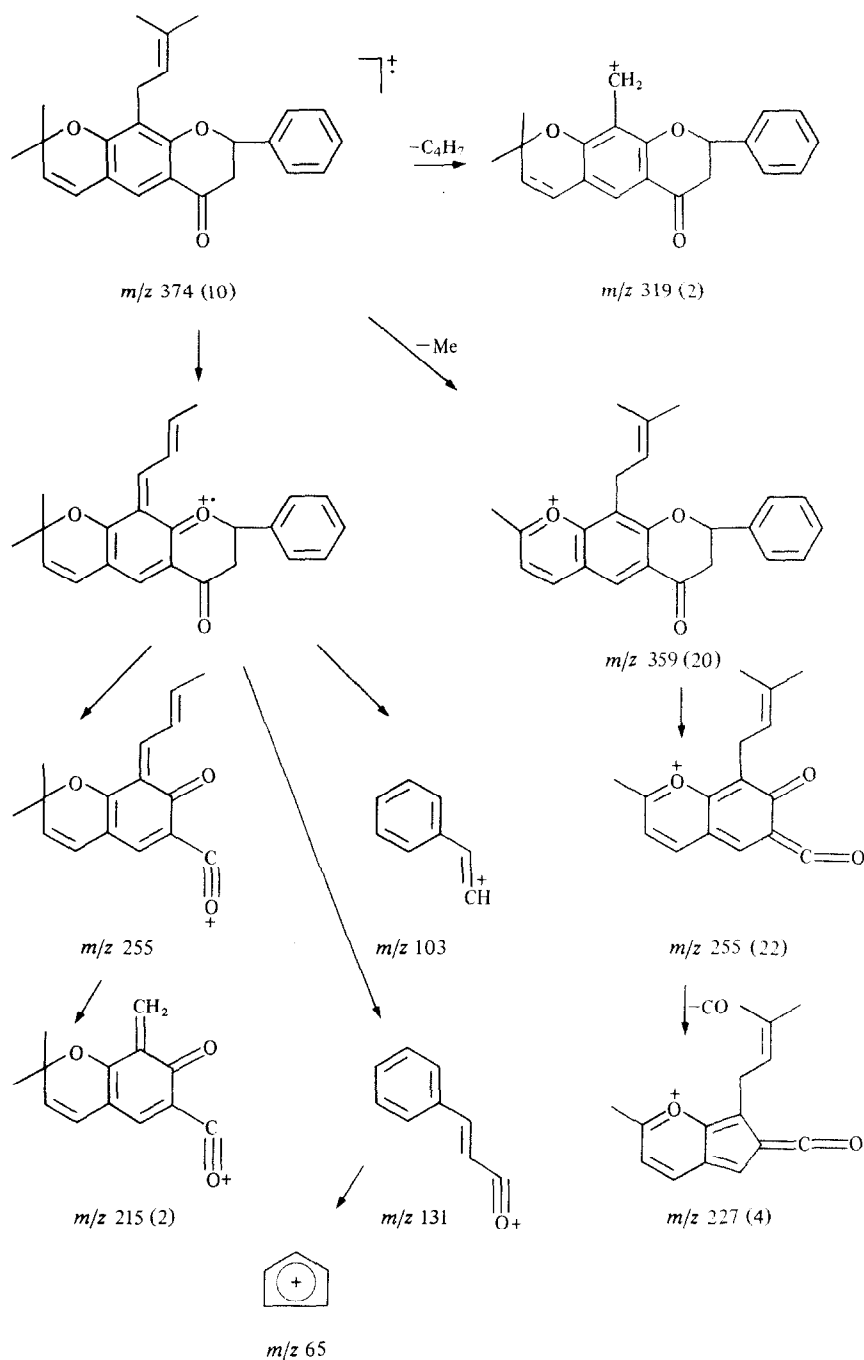
Extraction and isolation. Air dried leaves were extracted with Me_2CO under reflux. After concn under red. pres., the residue was macerated successively with hot hexane, C_6H_6 and EtOAc. The EtOAc extract was passed through a Sephadex LH 20 column for purification, the C_6H_6 -EtOAc eluates were then chromatographed on a silica gel 60 column without activation. The C_6H_6 -EtOAc (3:1) eluates yielded LA-1 (60 mg), mp 91–92°. 6,7,2,2-Dimethylchromeno-8, γ,γ -dimethylallyl flavanone, compound LA-1: UV λ_{max}^{EtOH} nm: 222, 272; MS m/z : 374 $[M]^+$ (10%), 359 (20), 319 (2), 255 (22), 227 (4), 55 (100); 1H NMR: δ 1.46 (6H, s, $>CMe_2$), 1.75, 1.8 ($2 \times 3H$, s, $>CMe_2$), 3.2

$\begin{matrix} H \\ | \\ (2H, d, J = 7 \text{ Hz}, -CH_2-) \end{matrix}$, $\begin{matrix} H \\ | \\ 5.4 (1H, t, =C-) \end{matrix}$, $\begin{matrix} H \\ | \\ 5.6, 6.7 (2 \times 1H, d, J = 11 \text{ Hz}, >C=O) \end{matrix}$, δ 7.5 (5H, s, unsubstituted phenyl), 7.7 (1H, s, 5-H), 3.0 (2H, m, H-3), δ 5.15 (1H, m, H-2).

Acknowledgement—The authors are greatly indebted to Dr. M. R. Parthasarthy, Delhi University, Delhi, India for his constructive suggestions during the course of this study.

REFERENCES

1. Oliver, D. (1968) *Flora of Tropical Africa*, Vol. 1, p. 446.
2. Seikel, M. K. and Geissman, T. A. (1950) *J. Am. Chem. Soc.* **72**, 574.



Scheme 1. Mass spectral fragmentation of the flavanone from *Lannea acida*.

- Jurd, L. (1962) *The Chemistry of Flavonoid Compounds* (Geissman, T. A., ed.) pp. 107–155. Pergamon Press, Oxford.
- Burrows, B. F. and Ollis, W. D. (1960) *Proc. Chem. Soc.* 177.
- Pelter, A. and Stainton, P. (1966) *J. Chem. Soc. (C)* 701.
- Wolform, M. L. and Wildi, B. (1951) *J. Am. Chem. Soc.* **73**, 235.
- Richards, J. H. and Hendrickson, J. B. (1966) *The Biosynthesis of Steroids, Terpenes and Acetogenins*, pp. 27–172. W. A. Benjamin, New York.